Europium Arsenic Oxide Eu_4As_2O : A Filled La_2Sb Structure and its Relation to the K_2NiF_4 and GeTeU Types*

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Eu₄As₂O, tetragonal, 14/mmm (D_{4h}^{17}) ; Z = 2, FW 773.68, a = 4.7924 (4), c = 16.1933 (9) Å, c/a = 3.379, U = 371.9 Å³; F(000) = 649, $D_x = 6.71$ g cm⁻³; $\mu = 439$ cm⁻¹ (Mo Ka). Diffraction data from a crystal $\sim 0.03 \times 0.07 \times 0.03$ mm with Mo Ka radiation and a graphite monochromator gave $R_2 = 3.8\%$ for 529 observed *hkl* when corrected for absorption and refined by full-matrix least squares with allowance for anomalous dispersion. The structure is a filled version of the La₂Sb type with O atoms occupying the octahedral holes at 2(b) $(0,0,\frac{1}{2})$, As atoms in 4(e) |z = 0.1354 (2)| and Eu atoms in 4(e) with z = 0.3261 (1) and 4(c) $(0,\frac{1}{2},0)$. The average O–Eu distance is 2.54 Å but the individual distances O–Eu(1) = 2.40 (×4) and O–Eu(2) = 2.82 Å (×2) may indicate the different nature of the two independent Eu atoms. The As atom has nine Eu atoms as nearest neighbors with an average As–Eu of 3.35 Å; this arrangement is usually described in terms of a tricapped trigonal prism but the description in terms of a monocapped square antiprism is more convenient in this case because there is an As contact of 4.39 A opposite the uncapped square face. Both Eu atoms can be described as 14-coordinated with Eu(1) clearly occupying less volume; average distances for CN = 14 are Eu(1) = 3.30 and Eu(2) = 3.66 Å. This structure is closely related to the K₂NiF₄ and GeTeU types.

Introduction

The structure analysis of Eu₄As₂O was undertaken to resolve a difficulty encountered in a study of the binary EuAs system (Taylor, Calvert, Utsonomiya & Wang, 1977). Samples containing less than 42 at.% As which were slowly cooled from above the liquidus in sealed Ta tubes were frequently found to contain minor amounts of a phase which could not be placed in the phase diagram. This phase was not obtained as a major component, and was also found in unmelted samples containing 38-42 at.% As, prepared in silica, Ta and Mo crucibles, if the preparation temperature exceeded 1100°C. It was found as well-formed crystals in arcmelted samples with nominal compositions in the range 39–41 at.% As. Although it was suspected prior to the structure analysis that this was a ternary phase, no conclusive proof of its nature was obtained from chemical analysis or by synthetic methods, because it was never obtained as a single phase. The total overall O and Ta content of the samples was ~ 1000 p.p.m.a. (parts per million atomic) and 30 p.p.m.a. respectively as compared with 1200 p.p.m.a. O and 10 p.p.m.a. Ta in the Eu used in preparing the samples. The best crystals were obtained from a sample, with nominal composition ~ 38 at.% As, cooled from the liquidus (~1700°C) in a sealed Ta crucible. Debye-Scherrer

and Guinier photographs confirmed that these crystals were of the desired phase and precession photographs yielded the diffraction symbol 4/mmm. I^{***} (h + k + l)= 2n for all observed reflections) with cell parameters which could be used to index the Guinier photographs. The crystals were sealed in thin-walled glass capillaries in an argon-atmosphere dry box which was used for all specimen preparation operations (Ono, Hui, Despault, Calvert & Taylor, 1971). A well-formed crystal which gave satisfactory Laue photographs was used to collect intensity data on a four-circle computer-controlled diffractometer with local programs (Wang, Gabe, Calvert & Taylor, 1976). The lattice parameters, obtained by centering 30 *hkl* reflections with $2\theta > 70^{\circ}$, were a = 4.7927 (5), b = 4.7920 (4), c = 16.1932 (9) Å, $\alpha = 89.969$ (6), $\beta = 89.959$ (6) and $\gamma = 89.976$ (7)° when refined as triclinic. The values used in the structure refinement are those given in the Abstract. Intensities were measured with a θ -2 θ scan technique [width = $(0.7 + 0.7 \tan \theta + 0.7)^\circ$; background measured for a time equal to scan time]; the 1264 accessible reflections with $2\theta < 100^{\circ}$ were measured. These measurements were averaged over 632 symmetry-related pairs, corrected for Lorentz and polarization factors and absorption effects yielding 529 independent *hkl* with $I > 2\sigma$ (84% observed). The internal consistency between symmetry-related sets, $\Sigma |\Delta I| / \Sigma I$, was 0.06. The diffraction data indicated that the structure was of the La₂Sb type although the nominal composition was ~Eu_sAs₃. A full-matrix least-

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Table 1. Atomic parameters for $Eu_4As_2O(B_{ij} \times 10^4)$

	x	y z	β_{11}^*	β_{22}	β_{33}	B _{iso} †
Eu(1) in $4(c)$	0	± 0	30 (2)	104 (3)	13 (1)	0.91 (2)
$E_{u}(2)$ in $4(e)$	0	Ó 0-3261 (1)	149 (2)	149 (2)	8(1)	1.19 (3)
As in $4(e)$	0	0 0.1356 (1)	91 (3)	91 (3)	6(1)	0.81 (5)
O in $2(b)$	0	$0\frac{1}{2}$	40 (25)	40 (25)	42 (7)	0.41 (-

* $T = \exp[-(h^2\beta_{11}^2 + \dots + hk\beta_{12} \dots)]; \beta_{12} = \beta_{23} = \beta_{13} = 0.$ † $B_{iso} = equivalent isotropic B; T = \exp[-B(\sin \theta/\lambda)^2].$

Table 2. Interatomic distances in Eu₄As₂O (Å)

Eu(1)–O 2.3962	2 (1) (×2)	Eu(2)–O	2.8162	2 (6)
-As 3.2499	9 (8) (×4)	-As	3.085	(1)
-Eu(1) 3.3887	7 (1) (×4)	-As	3.445	l (3) (×4)
-Eu(2) 3.697	7 (5) (×4)	-Eu(1)	3.697	7 (5) (×4)
As-As 4-391	(2) (×4)	-Eu(2)	4.1900) (7) (×4)
Average values				
Eu(1)–Eu	3.54	Eu(2)	-Eu	3.93
Eu(1)–As	3.25	Eu(2)	-As	3.37
Eu(1)-all	3.30	Eu(2)	-all	3.66
As-Eu	3.35	O-E	L	2.54

Table 3. Guinier focusing-camera pattern of Eu₄As₂O

Diameter = 80 mm, Cu $K\alpha_1$, $\lambda\alpha_1 = 1.54060$ Å, 22°C, silicon standard a = 5.43052 Å, d_c from refined parameters a = 4.7939 (7), c = 16.1936 (24) Å gave $M_{20} = 70$ (de Wolff, 1968). I_c are integrated intensities based on the parameters of Table 1.

$d_{o}(\mathbf{\dot{A}})$	d_{c} (Å)	h k l	I_o	I _c
8.11	8.10	002	w	17
3.586	3.586	103	w	13
3.390	3.390	110	w	10
3.127	3.127	112	vs	100
2.700	2.699	006	m	33
2.684	2.684	105	m	25
2.600	2.599	114	vs	77
2.398	2.397	200	5	63
1.792	1.792	206	т	35
	1.788	215		13
1.737	1.738	118	w	10
1.695	1.695	220	m	19
1.690	1.685	109	w	5
1.4903	1.4901	312	w	18
1.4612	1.4612	1 1.1 0	w	6
1.436	1.4353	226	m	15
1.4191	1.4197	314	m	19

squares refinement starting from the La₂Sb coordinates (Stassen, Sato & Calvert, 1970) converged to $R_2 =$ 0.11. A difference Fourier map showed anisotropic thermal motion and revealed a large peak at $(0,0,\frac{1}{2})$. Continued refinement with anisotropic thermal parameters and allowance for anomalous dispersion gave $R_2 = 0.069$ but the large peak at $(0,0,\frac{1}{2})$ remained. Refinements with partial occupancy of this site by Ta, Eu or As were unsuccessful. However, a refinement



Fig. 1. Atomic coordinations in Eu₄As₂O. (a) O atom at $(\frac{1}{2},\frac{1}{2},0)$ [2(b)]. (b) Eu(1) atom at $(0,\frac{1}{2},0)$ [4(c)]. (c) Eu(2) atom at $(\frac{1}{2},\frac{1}{2},0\cdot1379)$ [4(e)]. (d) As atom at $(0,0,0\cdot1356)$ [4(e)]. The origin is indicated in each case; O is represented by small solid circles, Eu(1) by stippled circles, Eu(2) by open circles and As by hatched circles.

with full occupancy by O converged and showed no significant detail in the final difference Fourier map. The parameters, interatomic distances and an indexed powder pattern are given in Tables 1, and 2 and 3.* The final agreement factors are $R_1 = \Sigma |\Delta F| / \Sigma F_o = 0.050$, $R_2 = [\omega \Delta F^2 / \omega F_o^2]^{1/2} = 0.038$, goodness of fit $[\Sigma \omega (\Delta F)^2 / (N - V)]^{1/2} = 1.37$ where N = 529 and V = 11 (Wang *et al.*, 1976). All calculations were carried out on the laboratory PDP-8e computer with local programs. For reasons given below a refinement based

^{*} A list of structure factors has been deposited with the British Library Lending Division as Supplementary Publication No. SUP 32600 (6 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 13 White Friars, Chester CH1 1NZ, England.

on the K₂NiF₄ arrangement was tried but was found to be unsuccessful. It should be noted that the X-ray refinement does not exclude the possibility that the 2(b)site is occupied by N (or by N and O in a disordered manner); the chemical analysis shows about ten times as much O as N in the samples and this supports the occupancy by O atoms. Subsequent to this structure analysis, microprobe analyses were carried out in the laboratories of the Department of Energy, Mines and Resources. Attempts to analyze the actual crystal used failed, but six grains in two ingots which contained this phase were analyzed. With EuAs and Eu_sAs, used as standards, the analysis sums for Eu and As were less than 100% – if the difference is taken to be O the average values for each ingot are $Eu_{4\cdot00}As_{2\cdot00}O_{0\cdot84}$ and $Eu_{4\cdot01}As_{1\cdot99}O_{1\cdot34}$. This is consistent with, and indeed supports the structure analysis.

Description

The O at $(0,0,\frac{1}{2})$ (Fig. 1*a*) is bonded to four Eu(1) and two Eu(2) which form a distorted octahedron (Table 2). Eu(1) has CN = 14 with its neighbors forming an irregular polyhedron (Fig. 1*b*) with 12 vertices, with the two O lying along two of the sides of the square formed by Eu(1) atoms. Eu(2) is also 14-coordinated with As and O as strongly bonded neighbors (3.09 and 2.82 Å) and four Eu(2) at 4.19 Å with weak interactions. This coordination polyhedron is an irregular figure (Fig. 1*c*) having 13 vertices with the O lying at the center of the square face; this polyhedron can be derived by taking a square antiprism, formed by the four Eu(1) and four Eu(2), placing four As alternately opposite four of its eight

Table	4.	Structures	in	I4/mmm	closely	related	to		
Eu ₄ As ₂ O									

Average distances (Å)*							
		4(<i>e</i>)	4(<i>e</i>)	4(c)	z_1	z_2	
c/a	Formula	\boldsymbol{z}_1	<i>z</i> ₂	-	(×10 ³)	(×10 ³)	R(%)
3.26	F ₄ K ₂ Ni	2.59	2.98	2.52	151	352	11
3.36	Ca ₄ As ₂	3.16	3.60	3.31	135	328	
3.37	Eu₄As ₂ O	3.32	3.66	3.30	136	326	3.8
3.48	Sr ₄ Sb ₂	3.47	3.93	3.63	137	328	9.5
3.49	Ca ₄ Sb ₂	3.25	3.69	3.38	139	334	15
3.49	Ca ₄ Sb ₂	3.29†	3.94†	3.40		_	
3.51	Ca₄Bi,	3.28	3.74	3.42	140	334	15
3.55	Ba ₄ Bi ₂	3.67	4.17	3.85	137	327	7
3.90	La₄Sb,	3.35	3.78	3.55	138	320	14
4.15	P,Ťe,Ū,	3.04	3.50	2.52	115	310	11
4.16	As,Te,U,	3.07	3.55	2.93	120	315	6
4.28	Ge, Te, U,	3.08	3.53	2.94	123	312	5
3.58	Ti ₂ Bi ₂ Ti ₂	2.87	2.93	2.86	133	353	

* Average of all distances from central atom to its neighbors.

[†] Average values over two independent Ca atoms; this structure is refined in *I4mm* (Hamon *et al.*, 1975).

triangular faces and then adding an As, capping the Eu(2) square face, and an O in the center of the other square face. The As have nine near neighbors forming the tri-capped trigonal prism common among transition-metal pnictides and chalcogenides; the alternative description as a mono-capped square antiprism (Fig. 1d) is convenient here because there is a tenth neighbor (As at 4.39 Å) at what amounts to a van der Waals contact opposite the remaining square face.

Discussion

The La₃Sb structure type has, since its first description (Stassen et al., 1970), been found and described for a number of other phases: Ba₂Bi (Martinez-Ripoll, Haase & Brauer, 1974), Sr₂Sb (Martinez-Ripoll & Brauer, 1973), Ca,Sb and Ca,Bi (Eisenmann & Schäfer, 1974; Hamon, Marchand, L'Haridon & Laurent, 1975) and Ca₂As (Hutz & Nagorsen, 1974) so that no detailed geometrical description is necessary. One feature will be discussed here. These structures (Table 4) are all characterized by c/a values between 3.3 and 3.9 and z_1 and z_2 values $\simeq 0.135$ and 0.325respectively, values essentially identical with those of Eu₄As₂O. Thus, they all have two kinds of metal-atom environment, M(1) and M(2) with M(1) on the 4(c)site occupying much less volume. For Eu₄As₂O the domain volumes (Frank & Kasper, 1958) unique to Eu(1) and Eu(2) are, respectively, 22.5 and 33.3 Å³ with the corresponding average interatomic distances 3.30 and 3.66 Å. To explain this difference in the case of Eu₄As₂O, two arguments may be advanced. On the one hand Eu(1) may be smaller because of the tight bonding to O [Eu(1)–O(×2) = 2.40 Å], but if this is so, then a different explanation would be required for the other phases of this type in Table 4. However, these phases may contain O which was not detected in the structure analysis; we have canvassed this possibility for the La₂Sb data used by Stassen et al. (1970) and conclude that their photographic data would not be accurate enough to show the presence of O and do not warrant reprocessing and refining. For the other compounds of this type in Table 3 a study of the Rvalues and the refinements does not rule out the possibility that O was present but not detected; because in particular, no difference Fourier maps were calculated and the elements involved are either heavy or reactive or both. It therefore seems desirable to reexamine one of these structures again and we plan to do this. On the other hand, the difference between M(1) and M(2) for the Eu₄As₂O case may be ascribed to the presence of Eu³⁺ ($r \simeq 1.80$ Å) and Eu²⁺ ($r \simeq 2.04$ Å). Both the difference ($\Delta = 0.36$ Å) between the average M(1) and M(2)-neighbor distances and their absolute values, Eu(1)-X = 3.30, Eu(2)-X = 3.66 Å, are compatible with this explanation, but for the other phases of this

type in Table 4 no such variable valency is known to occur. Nevertheless the average As-Eu distance (3.35 Å) is closer to that predicted for Eu²⁺-As (3.43 Å) than that predicted for Eu³⁺-As (3.19 Å) (based on CN = 12 radii) suggesting the nominal valence formulation $Eu_4^{2+}As_2^{3-}O^{2-}$, consistent with divalency for all compounds in the second group of Table 4 and with the first explanation given above. Detailed discussion is more appropriate after one of these structures has been reexamined.

Related structure types

Table 4 lists selected data for other structures closely related to that of Eu₄As₂O. Only one representative of the K₂NiF₄ type is listed as this structure is well known (Muller & Roy, 1974; Baltz & Plieth, 1954, 1955). The UGeTe structure type was first reported and described by Haneveld & Jellinek (1969) and UPTe and UAsTe were described by Zygmunt, Murasik, Ligenza & Leciejewicz (1974). Although all structures are similar in a formal geometrical sense, the z_1 and z_2 coordinates and the distribution of atoms among the site sets clearly divide them into groups. The $K_2 NiF_4$ structure has F(1)and K atoms in the M(1) 4(c) and $M(2) 4(e) (z \sim 0.35)$ sites of the La₂Sb type whereas the UGeTe type has U in the trigonal prismatic site 4(e) ($z \sim 0.1$) occupied by the Group V element in the La₂Sb type and Ge and Te occupying the M(1) and M(2) sites respectively. Their relation to the parent type Cu₂Sb (PbFCl, SiZrSe) has been discussed by Haneveld & Jellinek (1969) and also by Zygmunt et al. (1974) and for the La₂Sb type this comparison has been discussed by Stassen et al. (1970). In Table 4, Ti₂Bi (Auer-Welsbach, Nowotny & Kohl, 1958) appears to hold an anomalous position although closely similar to K₂NiF₄. It has been discussed by Haneveld & Jellinek (1969) and Stassen et al. (1970) but the structure may deserve further study.

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